Melanin Synthesis Inhibitors from Lespedeza floribunda

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In the course of our search for new melanin synthesis inhibitors from plants, 40 new flavonoids and 11 known flavonoids were isolated from the roots of *Lespedeza floribunda* Bunge. The structures of the new compounds were determined by MS and NMR analyses, and the absolute configurations by CD spectra. Many of the compounds inhibited melanin synthesis in normal human epidermal melanocytes (NHEM), and compounds **3**, **7**, **8**, **11**, **16**, **24**, **27**, **29**, **33**, **43**, **45**, and **51** were particularly inhibitory. Their activities were stronger than that of hydroquinone, which is known as a major skin-lightening drug.

Skin pigmentations, such as melasma, freckles, and solar lentigo, can be serious aesthetic problems. They result from increased production and accumulation of melanin. Consequently, inhibition of melanin synthesis is very important for the treatment of a variety skin pigmentations. Melanin synthesis inhibitors have been of interest as target molecules of natural product chemistry. Prenyl flavanones from aerial parts of Lespedeza buergeri Miq. (Leguminosae) showed strong inhibition of melanin synthesis in normal human epidermal melanocytes (NHEM).¹ In the course of our search for active compounds from natural sources, we examined the diethyl ether layer of a methanol extract of roots of L. floribunda Bunge and isolated 51 flavonoids. Among these, 1-40 were new compounds and 11 were identified as bavachin (coryfolin) (41),² 6,8-di(γ , γ -dimethylallyl)-4',7-dihydroxyflavanone (42),³ prostratol (43),⁴ 3',6-di(γ , γ -dimethylallyl)-4',5,7-trihydroxyflavanone (44),⁵ euchrenone a_3 (45),⁶ cajaflavanone (erythrisenegalone) (46),⁷ genistein (47),⁸ erythrabyssin II (48),⁹ erybraedin A (49),¹⁰ ery-braedin B (50),¹⁰ and folitenol (51)¹¹ by comparison of reported NMR data. The inhibitory activities of these compounds on melanin synthesis in NHEM were also determined. The inhibitory activities of compounds 3, 7, 8, 11, 16, 24, 27, 29, 33, 43, 45, and 51 were all greater than that of hydroquinone, which is a major skinlightening drug.

Results and Discussion

The ¹H NMR spectra of compounds 1-4, named lespeflorins A1, A2, A3, and A4, showed a set of ABX-type proton signals in the aliphatic proton region, and the ¹³C NMR spectrum of these compounds each showed a carbonyl carbon signal, suggesting that they were flavanones.¹² In the ¹H NMR spectrum of **1**, a singlet aromatic proton signal at δ 7.61, a set of AA'XX' proton signals at δ 6.94 (2H, d, J = 8.5 Hz) and δ 7.38 (2H, d, J = 8.5 Hz), a methoxy singlet at δ 3.83, and two sets of isoprenyl proton signals at δ 3.30 (2H, d, J = 7 Hz), 3.40 (2H, d, J = 7 Hz) and δ 5.23 (t sept, J = 7, 1 Hz), 5.30 (t sept, J = 7, 1 Hz), 1.74 (3H, brs), 1.74 (3H, brs), 1.76 (3H, d, J = 1 Hz), 1.78 (3H, brs) were observed except for three proton signals for H-2 and H₂-3. In the ROE difference spectrum, an ROE was observed with the proton signal at δ 6.94 on irradiation at the methoxy signal, suggesting the presence of a 4'-OMe. In the HMBC spectrum, C-H long-range couplings were observed between the proton signal at δ 7.41 and the carbon signals at δ 29.1 (C- α of the isoprenyl at C-6), 114.5 (C-4a), 159.8 (C-8a), and 191.5 (C-4). The ¹H NMR spectrum of 2 was similar to that of 1, lacking one isoprenyl residue. In the aromatic proton region, a set of AX-type proton signals at δ 6.53, 7.75 (J = 8.5 Hz) and a set of AA'XX'-type proton signals at δ 6.95, 7.39 (J = 8.5 Hz) were observed. Lespeflorin A₃ (**3**) had a ¹H NMR spectrum similar to that of **1**. Three proton signals due to a 1,3,4-trisubstituted benzene ring were observed at δ 6.80 (d, J =8.5 Hz), 7.05 (d, J = 2.5 Hz), and 7.18 (dd, J = 8.5, 2.5 Hz). The proton signal at δ 7.05 was C-H long-range coupled with the carbon signals at δ 79.4 (C-2) and 122.0 in the HMBC spectrum. The latter carbon signal was assigned to an α -carbon of the dimethylchromene ring because of the chemical shift and coupling constant of the attached proton signal at δ 6.32 (d, J = 10 Hz). The ¹H NMR spectrum of **4** showed a methoxy singlet at δ 3.81, a hydrogen-bonded OH group at δ 12.14 (s), a 1,2,4-trisubstituted benzene ring [δ 6.45 (d, J = 2 Hz), 6.48 (dd, J = 8, 2 Hz), 7.40 (d, J = 8 Hz)], a pentasubstituted benzene ring [δ (5.99 (s)], and a dimethylchromene ring [δ 1.42 (3H, s), 1.44 (3H, s), 5.45 (d, J = 10 Hz), 6.54 (d, J = 10 Hz)]. The location of the OCH₃ group was decided to be 2' due to the ROE effect on H-2 and H-3' on irradiating at the signal at δ 3.81. The singlet aromatic proton at δ 5.99 was assigned to H-6 because that proton was attached to the carbon at δ 97.4, which was long-range coupled with the hydrogenbonded OH at δ 12.14 in the HMBC spectrum. The absolute configuration at C-2 of these flavanones was S because of their levorotatory optical activity¹³ and negative Cotton effects at the $\pi \rightarrow \pi^*$ absorption band.¹⁴ The structures of these four new compounds were determined as shown from the NMR, UV, and HRMS spectral data.

Lespeflorins B_1 (5), B_2 (6), B_3 (7), and B_4 (8) had a flavanonol skeleton showing typical AB-type proton signals in the aliphatic region of the ¹H NMR spectrum.¹⁵ The ¹H NMR spectrum of 5 had two singlet aromatic proton signals at δ 6.40 (s), 7.67 (s), a set of 1,3,4-trisubstituted benzene ring proton signals at δ 6.86 (d, J = 8 Hz), 7.28 (brs), 7.30 (dd, J = 8, 2 Hz), two sets of isoprenyl proton signals at δ 3.33 (2H, brd, J = 7 Hz), 3.39 (2H, brd, J = 7Hz) and 5.29 (t sept, J = 7, 1 Hz), 5.34 (t sept, J = 7, 1 Hz), 1.77 (9H, brs), 1.79 (3H, d, J = 1 Hz), and a set of AB-type proton signals at δ 4.55 (d, J = 12 Hz), 4.98 (d, J = 12 Hz) due to H-3 and H-2. In the HMBC spectrum, the proton signal at δ 7.67 was C-H long-range coupled with the carbon signals at δ 28.9 (C- α of isoprenyl at C-6) and 162.3 (C-7), 192.9 (C-4) and was assigned to H-5. Another proton singlet at δ 6.40 was assigned to H-8, as it showed C–H long-range couplings with the carbon signals at δ 112.5 (C-4a), 122.5 (C-6), 162.3 (C-7), 192.9 (C-4). The aromatic proton at δ 7.28 was C-H long-range coupled with the carbon signals at δ 29.9 (C- α of the isoprenyl at C-3'), 84.0 (C-2), 127.0 (C-6'), 155.3 (C-4'). Lespeflorin B₂ (6) had a ¹H NMR spectrum similar to that of 5. In the aromatic proton region, a proton singlet

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Chart 1



at δ 7.58 and a set of AA'XX'-type signals at δ 6.85, 7.41 (each 2H, d, J = 8.5 Hz) were observed, indicating two sets of isoprenyl proton signals. The singlet was assigned to H-5 from its chemical shift and the C–H long-range couplings with C-4 (δ 193.4), C-7 $(\delta 160.8)$, C-8a $(\delta 159.6)$, and C- α of the isoprenyl residue at C-6 (δ 29.0). In the ¹H NMR spectrum of lespeflorin B₃ (7), there were three sets of isoprenyl proton signals and a set of 1,3,4-trisubstituted benzene ring proton signals at δ 6.79 (d, J = 8.5 Hz), 7.27 (dd, J = 8.5, 2 Hz), 7.28 (brs), except for two aliphatic proton signals due to H-2 and H-3. The signal at δ 7.28 showed C–H long-range couplings with C-2 (\$\delta\$ 83.8), C-3' (\$\delta\$ 127.1), C-4' (\$\delta\$ 155.0) and then was assigned to H-2'. The¹H NMR spectrum of lespeflorin B₄ (8) was similar to that of 7 and showed the presence of a hydrogenbonded OH group (δ 11.49) instead of a singlet aromatic proton. The absolute configuration of these four new compounds was determined to be 2R, 3R from the large coupling constant (12-12.5)Hz) between H-2 and H-3,¹² a levorotatory optical activity,¹⁶ and a positive Cotton effect at the $n \rightarrow \pi^*$ absorption band in the CD spectrum.¹⁴ These data determined the structures of lespeflorins B_1-B_4 (5-8) to be as shown.

Lespeflorins C₁ (9), C₂ (10), C₃ (11), C₄ (12), C₅ (13), C₆ (14), and C₇ (15) had α -hydroxydihydrochalcone skeletons with one or two isoprenyl groups, as determined from their ¹H NMR spectra, showing ABX-type signals in the aliphatic region.¹⁷ The ¹H NMR spectrum of 9 showed resonances for the ABX-type protons at δ 2.82 (dd, J = 14, 8 Hz), 3.04 (dd, J = 14, 4.5 Hz), 5.15 (m), two singlet aromatic protons at δ 6.41 (s), 7.66 (s), a set of AA'XX'type aromatic protons at δ 6.74, 7.06 (each 2H, d, J = 8 Hz), an isoprenyl group at δ 3.27 (2H, brd, J = 7 Hz), 5.36 (brt, J = 7Hz), 1.72 (3H, brs), 1.77 (3H, brs), a hydrogen-bonded OH at δ

12.19 (s), an OH at δ 4.13 (brd, J = 7.5 Hz), and two phenolic OH protons at δ 8.04 (brs), 9.56 (s). In the HMBC spectrum, C-H long-range couplings were observed between the OH signal at δ 4.13 and the carbonyl carbon signal at δ 205.0, between the aromatic proton signal at δ 6.41 and the carbon signals at δ 111.5 (C-1'), 121.6 (C-5'), 163.9 (C-4'), 165.0 (C-2'), between the aromatic proton signal at δ 7.66 and the carbon signals at δ 111.5 (C-1'), 121.6 (C-5'), 163.9 (C-4'), 165.0 (C-2'), and between the aromatic proton signal at δ 7.06 and the carbon signals at δ 42.5 (C- β), 115.9 (C-3, C-5), 131.3 (C-2, C-6), 157.0 (C-4). Lespeflorins C₂ (10) and C₃ (11) showed NMR spectra similar to those of 9 except for methoxy proton signals at δ 3.78 (s) and 3.87 (s), respectively. The methoxy group was placed at C-4 in 10 and at C-2' in 11 from the ROE experiment. In the ¹H NMR spectrum of **12**, there were two methoxy singlets (δ 3.78 and 3.89) that were placed at C-2' and C-4 using ROE experiments. The ¹H NMR spectrum of 13 showed a methyl singlet at δ 2.14 instead of a singlet aromatic proton signal at δ 6.41 in the spectrum of 9. Long-range C–H correlations were observed between the methyl proton signal and the carbon signals at δ 112.0 (C-3'), 160.3 (C-4'), 162.1 (C-2'). The ¹H NMR spectrum of 14 showed two isoprenyl groups and lacked a singlet aromatic proton signal at δ 6.41, as in the spectrum of 9. Lespeflorin C₇ (15) showed a methoxy singlet at δ 3.91 and a set of dimethylchromeme ring proton signals at δ 1.46 (3H, s), 1.48 (3H, s), 5.57 (d, J = 10 Hz), 6.32 (d, J = 10 Hz) in the ¹H NMR spectrum. The methoxy group was located at C-2' from the ROE experiment, and the dimethylchromene ring was placed as shown from the C-H long-range correlations between the singlet aromatic proton at δ 7.60 and the carbon signals at δ 121.1 (C- α of the chromene ring), 159.7 (C-4'), 161.0 (C-2'), 200.0 (C=O). The absolute configuration





39

40

isoprenyl

isoprenyl

at C- α of these compounds was *S*, as the CD spectra showed a positive Cotton effect at 330–332 nm in **9**, **10**, **13**, and **14** (C-2' OH type) and a positive Cotton effect at 270–288 nm and a negative Cotton effect at 326–340 nm in **11**, **12**, and **15** (C-2' OMe type).¹⁸

The ¹H NMR spectrum of lespeflorin D₁ (**16**) showed a set of ABX-type proton signals at δ 4.50 (dd, J = 11, 11 Hz), 4.54 (dd, J = 11, 6 Hz), 4.29 (dd, J = 11, 6 Hz), which were characteristic of an isoflavanone,¹⁹ a methoxy proton signal at δ 3.72 (s), two AX-type signals in the aromatic proton region, and two set of isoprenyl proton signals. The aromatic proton signal at δ 6.83 (d, J = 8.5 Hz) was C–H long-range coupled with signals at δ 46.5 (C-3), 155.6 (C-4'), 157.5 (C-6'), and the aromatic proton at δ 7.79 was C–H long-range coupled with the carbons at δ 161.1 (C-7), 192.8 (C-4) in the HMBC spectrum. On irradiation at the methoxy proton signal, ROEs were observed at H-3 (δ 4.29), H- α (δ 3.42), H- β (δ 5.26). The CD spectrum of **16** showed a positive Cotton effect (+11389) at 325 nm (n $\rightarrow \pi^*$), suggesting the absolute configuration at C-3 to be R.¹⁴

The ¹H NMR spectrum of lespeflorin E₁ (**17**) showed signals at δ 5.05 (2H, d, J = 1.5 Hz) and 6.55 (1H, t, J = 1.5 Hz), which were characteristic of an isoflavene,²⁰ except for three methoxy signals and two aromatic AX-type proton signals. The lower field proton of the AX-type signal at δ 6.72 (δ , J = 8 Hz) showed C–H long-range couplings with signals at δ 121.9 (C-3), 124.9 (C-4a),

145.5 (C-8a), 149.3 (C-7), and another lower field proton at δ 6.94 (d, J = 8 Hz) showed C–H long-range couplings with signals at δ 110.8 (C-5'), 128.4 (C-3), 149.6 (C-4'). The ¹³C NMR chemical shifts (δ 60.5, 60.9, 61.0) suggested that the three OCH₃ groups were all hindered spatially.

Н

 CH_3

isoprenyl

Н

Н

н

Lespeflorin F_1 (18) also had three hindered OCH₃ groups (δ 59.8, 60.6, 61.0). In the ¹H NMR spectrum, a singlet olefinic proton signal (δ 7.11) was observed except for three OCH₃ and two AX-type proton signals in the lower field. The singlet proton was C-H longrange coupled with signals at δ 117.2 (C-1'), 125.1 (C-4a), 131.8 (C-8) in the HMBC spectrum and was assigned to H-3 of the 2-arylbebzofuran skelton.²⁰ The ¹H NMR spectrum of lespeflorin F_2 (19) showed a OCH₃ signal at δ 3.91 (s), a set of 1,2,4trisubstituted benzene ring proton signals at δ 5.65 (d, J = 2.5 Hz), 6.51 (dd, J = 8, 2.5 Hz), 7.35 (d, J = 8 Hz), a singlet aromatic proton signal at δ 7.12 (s), an isolated methylene signal at δ 4.70 (2H, s), and a set of isoprenyl proton signals. A ROE was observed between the singlet aromatic proton and the OCH₃ signal. The C-H long-range correlations were observed between the isolated methylene proton and the carbons at δ 117.4 (C-4a), 121.0 (C-3), 150.2 (C-2), between the singlet aromatic proton and the carbons at δ 117.4 (C-4a), 143.2 (C-8a), 145.8 (C-6), 149.1 (C-7), and between the doublet aromatic proton at δ 7.35 (d, J = 8 Hz) and the carbons at δ 150.2 (C-2), 157.1 (C-2'), 160.4 (C-4').

The ¹H NMR spectra of lespeflorins G_{1-12} (20–31) showed signals characteristic of pterocarpans in the aliphatic region.¹² Lespeflorin G₁ (20) had a OCH₃ singlet at δ 3.86, two singlet aromatic proton signals at δ 6.50 (s), 6.67 (s), a set of AX-type aromatic proton signals (δ 6.55, 7.26), and a set of isoprenyl proton signals except for four aliphatic proton signals for H₂-6, H-6a, and H-11a. In the ROE experiment irradiating at the OCH₃ signal, a ROE was observed at the singlet aromatic proton signal at δ 6.79, which had C–H long-range couplings with the carbon signals at δ 40.3 (C-6a), 141.1 (C-8), 146.7 (C-9), 154.1 (C-10a) in the HMBC spectrum. The singlet aromatic proton signal at δ 6.50 had C-H long-range couplings with the carbon signals at δ 117.1 (C-6b), 141.1 (C-8), 146.7 (C-9), 154.1 (C-10), and the doublet proton signal at δ 7.26 had C-H long-range couplings with the carbon signals at δ 78.8 (C-11a), 153.9 (C-4a), 155.7 (C-3) in the HMBC spectrum. Lespeflorin G₂ (21) had a OCH₃ signal at δ 3.77 (s), three singlet aromatic protons (δ 6.39, 6.73, 7.22), and two sets of isoprenyl proton signals except for four aliphatic proton signals due to H₂-6, H-6a, H-11a. The singlet protons at δ 6.39, 6.73, and 7.22 were assigned to H-4, H-7, and H-1, respectively, from their chemical shifts and HMBC correlations. The OCH₃ group was placed at C-9 from the ROE effects between the proton at δ 6.73 and H₂-6 and H-6a and comparison of the ¹³C NMR data with those of 20. Lespeflorin G_3 (22) showed a ¹H NMR spectrum similar to that of **21**, with one more OCH₃ signal at δ 3.97 (s) instead of the singlet aromatic proton signal at δ 7.22. Lespeflorin G₄ (23) had NMR data similar to those of **21**, and a methyl group at δ 2.16 (s) was observed instead of a OCH₃ signal of 21. In the ROE experiment on irradiating this methyl signal, a ROE was observed at the aromatic proton signal at δ 6.85, which was assigned to H-7 because of the HMBC correlation to C-6a (δ 40.3) and C-10a (δ 156.5). The ¹H NMR spectrum of lespeflorin G_5 (24) was similar to that of 23, and a set of AX-type aromatic proton signals were observed at δ 6.55, 7.28 (d, J = 8,5 Hz) instead of two singlet aromatic proton signals. The ¹H NMR spectrum of lespeflorin G₆ (25) was similar to that of 21, and a set of 1,2,4-trisubstituted benzene ring proton signals were observed at δ 6.40 (d, J = 2.5Hz), 6.53 (dd, J = 8.5, 2.5 Hz), 7.37 (d, J = 8.5 Hz) instead of a set of isoprenyl proton signals. The ¹H NMR spectrum of lespeflorin $G_7(26)$ was similar to that of 25 and showed the same signal pattern. The location of the OCH₃ group was decided to be C-8 from the ROE experiment. Lespeflorin G_8 (27) showed NMR data similar to those of 25 and two OCH₃ signals at δ 3.77 (s), 3.79 (s). The OCH₃ groups were indicated to be at C-3 and C-9 from the ROE experiments. The NMR data of lespeflorin $G_9(28)$ were very similar to those of 27, having two OCH₃ signals at δ 3.79 (s), 3.85 (s). The OCH₃ groups were placed at C-3 and C-8 from the ROE experiments. The ¹H NMR spectrum pattern of lespeflorin G_{10} (29) was similar to that of 25 and 26. The signal at δ 2.26 (s) was assigned to the C-8 methyl group based on the ROE experiment. Lespeflorin G_{11} (30) had an isoprenyl group and a dimethylchromene ring. The locations of these residues were decided from the HMBC correlations. Lespeflorin G_{12} (31) had a OCH₃ group and a dimethylchromene ring. The locations of these residues were decided to be as shown from the ROE experiment and the HMBC correlations. The absolute configurations of these pterocarpans were all 6aR and 11aR based on their levorotatory optical activities²¹ and positive Cotton effects at 260-310 nm in the CD spectrum (see Experimental Section).¹⁴

Lespeflorins H₁ (**32**) and H₂ (**33**) showed a singlet proton signal (H₂-6) characteristic of pterocarpenes near δ 5.5 in the ¹H NMR spectrum.¹⁹ In **32**, a OCH₃ group (δ 3.80) and an isoprenyl group were attached to the pterocarpene skeleton at C-3 and C-10, respectively, from the ROE and HMBC data. In **33**, the isoprenyl group appeared as a dimethylchromene ring.

Lespeflorins I_1 (34), I_2 (35), and I_3 (36) were assumed to have coumestan skeletons from their characteristic carbon signals (δ

158–159).²⁰ The ¹H NMR spectrum of lespeflorin I₁ (**34**) showed four singlet aromatic proton signals and a set of isoprenyl proton signals. The aromatic protons were assigned to be H-4, H-10, H-7, and H-1, respectively, from their HMBC correlations. Lespeflorin I₂ (**35**) had three aromatic proton singlets (δ 6.87, 7.12, 7.49) and two sets of isoprenyl group signals. The isoprenyl groups were located at C-2 and C-10 based on HMBC correlations. Lespeflorin I₃ (**36**) had a OCH₃ group and an isoprenyl group. The ROE experiment and the HMBC correlations decided the position of these residues.

Lespeflorins J_{1-4} (37–40) were assumed to be dimeric flavonoids composed of a pterocarpan and a 2-arylbenzofuran from their molecular formulas and ¹³C NMR spectra.¹⁹ The ¹H NMR spectrum of 37 showed a characteristic pterocarpan spin system at δ 3.16 (dd, J = 11, 11 Hz), 3.37 (dddd, J = 11, 7, 5, 1 Hz), 3.95 (dd, J= 11, 5 Hz), 5.48 (d, J = 7 Hz), three singlet aromatic proton signals at δ 6.39, 6.97, 7.32, a doublet aromatic proton signal at δ 7.39 (d, J = 1 Hz), AX-type aromatic proton signals at δ 6.60, 7.18 (d, J = 8.5 Hz), a hydrogen-bonded OH signal at δ 13.42, and four sets of isoprenyl signals. The ROE experiment on irradiating at a doublet proton signal at δ 7.39 suggested that this proton was H-7, which was coupled with H-6a in the COSY spectrum, and irradiating the doublet aromatic proton signal at δ 7.18 suggested that this proton was H-1 of the pterocarpan skeleton. In the HMBC spectrum, the aromatic proton at δ 7.39 had a C–H long-range coupling with the carbonyl carbon at δ 196.4 and the carbon at δ 164.5, which was assigned to be the carbon with a hydrogen-bonded OH group in the HMBC spectrum. Lespeflorin J_2 (38) showed NMR data similar to those of 37, lacking a set of isoprenyl proton signals and adding a OCH₃ signal at δ 3.79 (s). In the ROE experiment, irradiating at the OCH₃ signal led to the conclusion that the B ring of the 2-arylbenzofuran skeleton was 2'-hydroxy-4'-methoxyphenyl. The ¹H NMR spectrum of **39** was similar to that of 37, showing two more aromatic proton singlets $(\delta 6.37, 7.19)$ instead of a set of AX-type signals. The aromatic proton signal at δ 7.19 had C-H long-range couplings with the carbon signals at δ 28.3 (C- α of isoprenyl at C-2), 80.4 (C-11a), 112.0 (C-11b), 155.6 (C-4a), 157.2 (C-3). In the ¹H NMR spectrum of 40, the substitution pattern of the A ring of the pterocarpan skeleton was the same as that of 39, and the substitution pattern of the B ring of 2-arylbenzofuran was the same as that of 38. The absolute configurations of the pterocarpan skeletons of these compounds were indicated to be 6aR, 11aR from the levorotatory optical activities.21

To evaluate the inhibitory activities on melanin synthesis of the 51 isolated flavonoids, we measured IC_{50} values of melanin synthesis in NHEM.²² Forty-six compounds showed inhibitory activity (Table 8), and none of them affected cell viability (data not shown). Compounds **3**, **7**, **8**, **11**, **16**, **24**, **27**, **29**, **33**, **43**, **45**, and **51**, which have more lipophilic structures than the other compounds, showed the greatest activity. The IC_{50} of each of these compounds was lower than 2 μ M, while that of hydroquinone, a positive control, was 2.2 μ M. Thus, these compounds were more potent than hydroquinone, a compound widely used as a skin-lightening agent, at least in cell cultures. We are currently investigating the mechanism of inhibitory activity of these compounds.

Experimental Section

General Experimental Procedures. Optical rotations were measured on a JASCO DIP-1000 digital polarimeter. UV spectra were measured in methanol on a Hitachi U-2010 spectrophotometer. Circular dichroism spectra were measured on a JASCO J-20A spectrometer. ¹H (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded on a JEOL JNM α -400 FT-NMR spectrometer, and chemical shifts are give in δ with TMS as an internal standard at 35 °C. Inverse-detected heteronuclear correlations were measured using HMQC (optimized for ¹J_{C-H} = 145 Hz) and HMBC (optimized for ⁿJ_{C-H} = 8 Hz) pulse sequences with a pulse field gradient. HRFABMS data were obtained on a JEOL JMS 700

Table 1. NMR Spectroscopic Data (400 MHz) for Compounds 1–4

	1 ^{<i>a</i>}		2^a		3 ^{<i>a</i>}		4 ^a	
position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
2	5.38 (dd, 13, 3)	79.3	5.41 (dd, 13, 3)	79.4	5.32 (dd, 13, 3)	79.4	5.69 (dd, 12.5, 3.5)	74.3
3	2.81 (dd, 17, 3.5)	44.2	2.82 (dd, 17, 3)	44.0	2.80 (dd, 17, 3)	44.1	2.85 (dd, 17, 3.5)	42.5
	2.98 (dd, 17, 13)		3.00 (dd, 17, 13)		2.97 (dd, 17, 13)		2.90 (dd, 17, 12.5)	
4		191.5		191.3		191.5		196.7
4a		114.8		115.1		114.5		103.0
5	7.61 (s)	125.7	7.75 (d, 8.5)	126.5	7.60 (s)	125.7		163.9
6		121.8	6.53 (d, 8.5)	110.6		121.8	5.99 (s)	97.4
7		159.3		161.3		159.8		162.1
8		114.5		114.5		114.7		101.7
8a		159.8		160.7		159.3		
1'		131.5		131.3		131.5		119.8
2'	7.38 (d, 8.5)	127.5	7.39 (d, 8.5)	127.5	7.05 (d, 2.5)	124.2		157.4
3'	6.94 (d, 8.5)	114.1	6.95 (d, 8.5)	114.1		121.2	6.45 (d, 2)	99.0
4'		159.9		159.8		153.2		157.0
5'	6.94 (d, 8.5)	114.1	6.95 (d, 8.5)	114.1	6.80 (d, 8.5)	116.4	6.48 (dd, 8, 2)	107.2
6'	7.38 (d, 8.5)	127.5	7.39 (d, 8.5)	127.5	7.18 (dd, 8.5, 2.5)	127.0	7.40 (d, 8)	127.4
α					6.32 (d, 10)	122.0		
β					5.64 (d, 10)	131.2		
γ						76.5		
δ					1.45 (s)	28.1		
OH at 5							12.14 (s)	
OMe								
at 2'							3.81 (s)	55.5
at 4'	3.83 (s)	55.3	3.84 (s)	55.4				

^a Measured in CDCl₃. Isoprenyl groups were observed; two in 1, one in 2, 4, and three in 3.

 Table 2. NMR Spectroscopic Data (400 MHz) for Compounds 5–8

	5 ^{<i>a</i>}		6 ^{<i>a</i>}		7 ^a		8 ^{<i>a</i>}	
	$\delta_{\rm H}~(J~{\rm in}~{\rm Hz})$	$\delta_{ m C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
position								
2	4.98 (d, 12)	84.0	4.98 (d, 12.5)	83.6	4.96 (d, 12)	83.8	4.94 (d, 12)	83.1
3	4.55 (d, 12)	73.2	4.50 (d, 12.5)	73.3	4.51 (d, 12)	73.2	4.50 (d, 12)	72.5
4		192.9		193.2		193.4		196.5
4a		112.2		111.7		111.7		100.5
5	7.67 (s)	128.6	7.58 (s)	125.8	7.58 (s)	125.7		158.8
6		122.5		122.8		122.6		107.8
7		162.3		160.8		160.7		163.2
8	6.40 (s)	103.9		114.9		114.9		107.0
8a		162.5		159.6		159.7		157.9
1'		129.0		129.0		128.8		128.8
2'	7.28 (brs)	129.5	7.41 (d, 8.5)	129.0	7.28 (brs)	129.2	7.27 (brs)	129.3
3'		127.2	6.85 (d, 8.5)	115.5		127.1		127.0
4'		155.3		156.3		155.0		155.1
5'	6.86 (d, 8)	116.0	6.85 (d, 8.5)	115.5	6.79 (d, 8.5)	115.7	6.86 (d, 8)	115.9
6'	7.30 (dd, 8, 2)	127.0	7.41 (d, 8.5)	129.0	7.27 (dd, 8.5, 2)	126.7	7.28 (dd, 8, 2)	126.8
OH at 5							11.49 (s)	
OH at 7			6.22 (brs)				6.43 (brs)	

^a Measured in CDCl₃. Isoprenyl groups were observed; two in 5, 6 and three in 7, 8.

mass spectrometer in a positive mode and on a JEOL JMS SX 102 mass spectrometer in a negative mode using a *m*-nitrobenzyl alcohol matrix. Preparative HPLC was performed on a JASCO 800 instrument.

Plant Material. Roots of *Lespedeza floribunda* Bunge were purchased from Maruzen Pharmaceuticals Co., Onomichi, Hiroshima, Japan. The plant was authenticated by Dr. Bing-Hui Che, Herbarium, South China Botanical Garden, The Chinese Academy of Sciences (IBSC), Guizhou Province, China, and a voucher specimen (7183) has been deposited at IBSC.

Extraction and Isolation. The roots of *L. floribunda* (500 g) were extracted twice with MeOH (2 L) under reflux for 2 h. The combined MeOH extract was concentrated under reduced pressure to give brown residue (38 g). The MeOH extract was suspended in hot H₂O (500 mL) and extracted with ether. The ether layer gave a brown residue (20.2 g) after evaporation of ether under reduced pressure. The H₂O layer was extracted with ethyl acetate to give a brown residue (1.15 g). The ether layer was subjected to silica gel (400 g) column chromatography (CC) using hexane—ethyl acetate (99:1–97:3) to give 13 fractions (1A–1M). Fraction 1C (461 mg) was subjected to preparative HPLC [Inertsil ODS-3 3×50 cm; H₂O–CH₃CN (35: 65–15:85) linear gradient, UV, 280 nm] to give 1 (7.4 mg), **3** (6.5 mg), **24** (17.0 mg), **28** (7.7 mg), **31** (13.6 mg), **33** (68.5 mg), **49** (2.6

mg), and 50 (12.7 mg). Fraction 1D (2924 mg), preparative HPLC (PHPLC) [TSKgel ODS-80TS 5.5 \times 120 cm; solvent, H₂O-CH₃CN (35:65-19:81) linear gradient, UV, 280 nm] gave 26 fractions (3A-3Z). Fraction 1E (652 mg) [PHPLC, Inertsil ODS-3 3 \times 50 cm; solvent, H₂O-CH₃CN (40:60-30:70) linear gradient, UV, 280 nm] gave 7 (14.0 mg), 29 (6.2 mg), 30 (11.6 mg), and four fractions (4A-4D). Fraction 1F (620 mg) [PHPLC, column, Inertsil ODS-3 3 \times 50 cm; solvent, H₂O-CH₃CN (40:60-25:75) linear gradient, UV, 280 nm] yielded 2 (2.3 mg), 4 (1.7 mg), 7 (40.6 mg), 8 (2.9 mg), 10 (39.1 mg), 17 (31.2 mg), 21 (13.5 mg), 22 (15.5 mg), 26 (1.8 mg), 29 (4.2 mg), 30 (7.2 mg), 42 (0.7 mg), 44 (2.4 g), 46 (1.4 mg), 48 (2.1 mg), and one fraction (5A). Fraction 1G (2078 mg) [PHPLC, column, TSKgel ODS-80TS 5.5 \times 120 cm; solvent, H₂O-CH₃CN (40:60-8: 92) linear gradient, UV, 280 nm] gave 21 (46.1 mg) and two fractions (6A and 6B). Fraction 1H (2114 mg) was subjected to silica gel (70 g) (silica gel PSQ-100B) CC using hexane-ethyl acetate (75:25) to give fractions 7A and 7B. Fraction 1J (3578 mg) [PHPLC, TSKgel ODS-80TS 5.5 \times 120 cm; solvent, H₂O-CH₃CN (35:65-19:81) linear gradient, UV, 280 nm] gave 35 (82.5 mg), 47 (26.5 mg), and three fractions (8A-8E). Fraction 1K (990 mg) [PHPLC, TSKgel ODS-80TS 5.5×120 cm; solvent, H₂O-CH₃CN (40:60 - 24:76) linear gradient, UV, 280 nm] yielded 19 (9.1 mg) and 34 (9.7 mg). Fraction 3C (49.9

	•6		10^a		11 ^a		12 ^{<i>a</i>}		13^a		14^a		15 ^a	
position	$\delta_{\rm H} (J \text{ in Hz})$	δ_{C}	$\delta_{\rm H} (J \text{ in Hz})$	δ_{c}	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	δ_{c}	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
-		129.3		128.5		129.8		130.1		128.4		128.7		128.8
2	7.06 (d, 8)	131.3	7.06 (d, 8.5)	130.4	7.01 (d, 8.5)	130.5	7.08 (d, 8.5)	130.4	6.99 (d, 8.5)	130.5	7.01 (d, 8.5)	130.5	7.03 (d, 8)	130.5
3	6.74 (d, 8)	115.9	6.81 (d, 8.5)	114.0	6.68 (d, 8.5)	115.1	6.80 (d, 8.5)	113.7	6.70 (d, 8.5)	115.3	6.71 (d, 8.5)	115.5	6.70 (d, 8)	115.1
4		157.0		158.6		154.3		158.3		154.6		155.9		154.3
5	6.74 (d, 8)	115.9	6.81 (d, 8.5)	114.0	6.68 (d, 8.5)	115.1	6.80 (d, 8.5)	113.7	6.70 (d, 8.5)	115.3	6.71 (d 8.5)	115.5	6.70 (d, 8)	115.1
9	7.06 (d, 8)	131.3	7.06 (d, 8.5)	130.4	7.01 (d, 8.5)	130.5	7.08 (d, 8.5)	130.4	6.99 (d, 8.5)	130.5	7.01 (d, 8.5)	130.5	7.03 (d, 8)	130.5
α	5.15 (m)	74.2	5.17 (ddd, 8, 7, 4.5)	72.9	5.31 (m)	77.2	5.26 (ddd, 7.5, 7, 3.5)	77.2	5.20 (brdd, 7, 4)	72.7	5.18 (ddd, 7.5, 7, 4)	72.9	5.27 (m)	77.2
β	2.82 (dd, 14, 8)	42.5	2.88 (dd, 14, 7)	42.4	2.62 (dd, 14, 7.5)	40.2	2.63 (dd, 14, 7.5)	40.3	2.86 (dd, 14, 7)	42.4	2.81 (dd, 14.5, 7)	42.5	2.62 (dd, 14, 7.5)	40.3
β	3.04 (dd, 14, 4.5)		3.10 (dd, 14, 4.5)		3.07 (dd, 14, 3.5)		3.07 (dd, 14, 3.5)		3.10 (dd, 14, 4)		3.10 (dd, 14.5, 4)		3.07 (dd, 14, 3)	
C=0		205.0		202.9		200.2		200.3		202.9		203.0		200.0
1'		111.5		110.8		116.6		116.8		110.0		110.1		116.7
2,		165.0		164.1		159.9		159.9		162.1		161.5		161.0
3,	6.41 (s)	103.5	6.38 (s)	104.0	6.44 (s)	9.66	6.44 (s)	9.66		112.0		114.4	6.41 (s)	99.8
, 4		163.9		162.3		161.1		161.0		160.3		160.9		159.7
5'		121.6		119.7		119.9		119.6		118.4		119.8		114.7
6,	7.66 (s)	132.1	7.33 (s)	130.7	7.70 (s)	133.6	7.68 (s)	133.7	7.24 (s)	127.7	7.29 (s)	128.0	7.60 (s)	130.0
λ		133.5		135.7		135.9		136.1		136.4		135.7		78.1
ò	1.77 (brs)	25.9	1.82 (brs)	25.8	1.79 (brs)	25.8	1.80 (s)	25.8	1.82 (brs)	25.8	$1.77 (d, 1)^b$	25.8	$1.48 (s)^d$	28.6
Me at 3'									2.14 (s)	7.5				
OMe at 4			3.78 (s)	55.2			3.78 (s)	55.2						
OMe at 2'					3.87 (s)	55.7	3.89 (s)	55.7					3.91 (s)	55.8
OH at 2'	12.19 (s)		11.87 (s)						12.22 (s)		12.27 (s)			
OH at 4'	9.56 (s)		6.28 (brs)		6.15 (brs)		5.94 (brs)		6.37 (s)		6.37 (s)			
OH at 4	8.04 (brs)													
OH at α	4.13 (brd, 7.5)		3.62 (brd, 8)		4.08 (brd, 6.5)		3.92 (d, 7)				3.68 (d, 7.5)			
^a Measure	ad in CDCl ₃ . Isoprei	nyl grou	ps were observed; one	in 9-13	3, 15 and two in 14.									

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 Table 4. NMR Spectroscopic Data (400 MHz) for Compounds

 16 and 17

	16 ^{<i>a</i>}		1 7 ^{<i>a</i>}	
	$\delta_{\rm H}~(J~{\rm in}~{\rm Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{ m C}$
position				
2	4.50 (dd, 11, 11)	71.9	5.05 (d, 1.5)	68.3
2	4.54 (dd, 11, 6)			
3	4.29 (dd, 11, 6)	46.5		128.4
4		192.8	6.55 (t, 1.5)	121.9
4a		115.7		124.9
5	7.79 (d, 8.5)	127.1	6.72 (d, 8)	121.9
6	6.52 (d, 8.5)	110.5	6.54 (d, 8)	107.7
7		161.1		149.3
8		114.4		134.8
8a		161.1		145.5
1'		120.9		117.8
2'		157.5		150.7
3'		120.7		139.9
4'		155.6		149.6
5'	6.58 (d, 8.5)	112.4	6.72 (d, 8)	110.8
6'	6.83 (d, 8.5)	127.8	6.96 (d, 8)	123.5
OMe at 2'	3.72 (s)	62.2	3.82 (s)	60.5
OMe at 3'			3.94 (s)	60.9
OMe at 8			3.97 (s)	61.0

^a Measured in CDCl₃. Two isoprenyl groups were observed in 16.

Table 5. NMR Spectroscopic Data (400 MHz) for Compounds18 and 19

	18 ^{<i>a</i>}		19 ^b	
position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
2		151.4		150.2
3	7.11 (s)	104.1		121.0
4			4.70 (s)	55.8
4a		125.1		117.4
5	7.11 (d, 8.5)	114.4	7.12 (s)	99.8
6	6.89 (d, 8.5)	111.6		145.8
7		144.6		149.1
8		131.8		111.9
8a		144.3		143.2
1'		117.2		110.7
2'		150.0		157.1
3'		140.4	5.65 (d, 2.5)	104.4
4'		149.8		160.4
5'	6.83 (d, 8.5)	110.9	6.51 (dd, 8, 2.5)	108.6
6'	7.58 (d, 8.5)	121.9	7.35 (d, 8)	132.5
OMe at 2'	3.95 (s)	59.8		
OMe at 3'	4.00 (s)	61.0		
OMe at 6			3.91 (s)	57.0
OMe at 8	4.30 (s)	60.6		

^{*a*} Measured in CDCl₃. ^{*b*} Measured in acetone-*d*₆. One isoprenyl group was observed in **19**.

mg) [preparative TLC (PTLC), silica gel PF₂₅₄; hexane-ethyl acetate (65:35)] gave 18 (22.6 mg). Fraction 3H (36 mg of 417 mg) [PTLC, hexane-ethyl acetate (65:35)] gave 27 (12.1 mg). Fraction 3O (23.0 mg) [PTLC, hexane-ethyl acetate (65:35)] gave 23 (11.5 mg). Fraction 3R (17.6 mg) [PTLC, hexane-ethyl acetate (7:3)] gave 51 (7.1 mg). Fraction 3T (50.4 mg) [PTLC, hexane-ethyl acetate (7:3)] gave 43 (34.7 mg). Fraction 3W (17.0 mg) [PTLC, hexane-ethyl acetate (7: 3)] yielded 45 (9.7 mg). Fraction 4B (22.2 mg) [PTLC, hexane-ethyl acetate (6:4)] gave 20 (11.0 mg). Fraction 4D (10.6 mg) [PTLC, hexane-ethyl acetate (7:3)] provided 33 (2.9 mg). Fraction 5A (39.1 mg) [PTLC, Develosil 60–10 2 \times 25 cm; hexane-ethyl acetate (75: 25), UV, 280 mn] gave 32 (7.4 mg). Fraction 6A (38.3 mg) [PTLC, hexane-ethyl acetate (7:3)] provided 10 (14.6 mg) and 12 (5.6 mg). Fraction 6B (68.9 mg) [PHPLC, Capcellpak phenyl 2 × 25 cm; H₂O-CH₃CN (42:58), UV, 280 nm] gave 14 (26.2 mg). Fraction 7A (562 mg) [PHPLC, Inertsil ODS-3 3 \times 50 cm; H₂O-CH₃OH (20:80), UV, 280 nm] yielded 5 (6.2 mg), 6 (10.4 mg), 16 (8.8 mg), 36 (12.0 mg), and 38 (23.1 mg). Fraction 7B (264 mg) [preparative HPLC, Inertsil ODS-3 3 × 50 cm; H₂O-CH₃CN (35:65-20:80), UV, 280 nm] provided 15 (6.6 mg), 36 (19.6 mg), and 40 (27.0 mg). Fraction 8A (37.6 mg) [PTLC, hexane-ethyl acetate (6:4)] gave 9 (14.3 mg). Fraction 8B (51.7 mg) [PTLC, hexane-ethyl acetate (55:45)] yielded

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	20 ^{<i>a</i>}		21 ^{<i>a</i>}		22^{a}		23^{a}		24^a	
position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\mathrm{H}} \left(J \text{ in Hz} \right)$	$\delta_{\rm C}$
$\frac{1}{2}$	7.26 (d, 8.5) 6.55 (d, 8.5)	129.3 109.9 155.7	7.22 (s)	132.0 120.9 155.5		159.7 114.2 157 1	7.24 (s)	132.0 120.9 155.6	7.28 (d, 8.5) 6.55 (d, 8.5)	129.4 109.7 155.6
4		115.0	6.39 (s)	103.9	6.28 (s)	100.4	6.39 (s)	103.9		114.9
6	3.62 (dd, 11, 11)	66.9	3.66 (dd, 11, 11)	66.2	3.63 (dd, 11, 11)	66.0	3.58 (dd, 11, 11)	66.7	3.59 (dd, 11, 11)	67.0
6 6a	4.27 (dd, 11, 5) 3.48 (ddd, 11, 7, 5)	40.3	4.21 (dd, 11, 5) 3.48 (ddd, 11, 7, 5)	40.8	4.17 (dd, 11, 5) 3.33 (m)	40.1	4.19 (dd, 11, 5) 3.45 (ddd, 11, 7, 5)	40.3	4.26 (dd, 11, 5) 3.46 (m)	40.3
11a 11b	5.46 (d, 7)	78.8	5.39 (d, 7)	112.8	5.57 (d, 6.5)	75.0 103.3	5.39 (d, 7)	112.6	5.42 (d, 7)	78.5
10b 7	6.79 (s)	$117.1 \\ 108.0$	6.73 (s)	122.0 108.8	6.75 (s)	122.2 108.9	6.85 (s)	118.0 123.6	6.84 (s)	118.0 123.6
8 9		141.1 146.7		143.2 145.3		143.1 145.3		116.4 153.6		116.3 153.6
10 10a	6.50 (s)	98.1 154.1		117.9 151.7		117.8 151.7		109.6 156.5		109.5 156.5
OMe-at 1 OMe-at 3					3.97 (s)	63.3				
OMe-at 8 OMe-at 9	3.86 (s)	57.3	3.77(s)	61.4	3 77 (s)	61.4				
Me at 8 OH at 3			5.32 (brs)	0111	5.17 (6)	0111	2.16 (s)	15.7	2.16 (s)	15.7
OH at 8 OH at 9			5.66 (brs)							
	25 ^a		26 ^a		27 ^a		28 ^a		20 ^{<i>a</i>}	
position	δ (Lin Hz)	δα	δ _{er} (Lin Hz)	δα	δ (Lin Hz)	δα	δ _{er} (Lin Hz)	δα	δ (Lin Hz)	δα
1	7.37 (d. 8.5)	132.3	7.40 (d. 8.5)	132.4	7.42 (d. 8.5)	132.0	7.43 (d. 8.5)	132.0	7.39 (d. 8.5)	132.4
2	6.53 (dd, 8.5, 2.5)	109.6	6.54 (dd, 8.5, 2.5)	109.6	6.63 (dd, 8.5, 2.5)	109.1	6.64 (dd, 8.5, 2.5)	109.0	6.54 (dd, 8.5, 2.5)	109.6
4	6.40 (d, 2.5)	103.5	6.40 (d 2.5)	103.8	6.47 (d, 2.5)	101.6	6.46 (d, 2.5)	100.9	6.41 (d, 2.5)	103.8
4a 6	3.69 (dd, 11, 11)	156.5 66.2	3.67 (dd. 11, 11)	156.7 66.6	3.71 (dd, 11, 11)	156.6 66.2	3.68 (dd, 11, 11)	156.6 66.6	3.62 (dd, 11, 11)	156.7 66.7
6	4.22 (dd, 11, 5)	40.6	4.22 (dd, 11, 5)	40.7	4.25 (dd, 11, 5)	40.8	4.24 (dd, 11, 5)	40.8	4.21 (dd, 11, 5)	40.3
11a	5.41 (d, 7)	77.4	5.49 (ddd, 11, 7, 5) 5.41 (d, 7)	77.2	5.44 (d, 7)	77.2	5.43 (d, 7)	77.5	5.41 (d, 7)	77.9
11b 10b		113.0 122.0		113.4 115.7		112.9 122.1		113.1 115.8		113.1 117.8
7 8	6.73 (s)	108.8 143.1	6.66 (s)	105.1	6.74 (s)	108.7	6.66 (s)	105.1	6.85 (s)	123.6
9		145.3		141.1		145.4		141.1		153.4
10 10a		118.0 151.6		111.8 152.5		118.0 151.7		111.8 152.5		109.6 156.4
OMe-at 1 OMe-at 3					3 79 (s)	55.4	3.79(s)	55 5		
OMe-at 8	2.76(c)	61.5	3.85 (s)	57.0	2 77 (c)	61.4	3.85 (s)	57.0		
Me at 8	5.70 (8)	01.5			5.77 (8)	01.4			2.16 (s)	15.7
OH at 3 OH at 8										
OH at 9			5.69 (brs)							
•,•	$\frac{30^a}{2}$								<u>34^a</u>	
position	$O_{\rm H}$ (J in Hz)	132.0	$O_{\rm H} (J \text{ in Hz})$	131.9	$O_{\rm H} (J \text{ in Hz})$	0 _C	$O_{\rm H} (J \text{ in Hz})$	0 _C	$O_{\rm H}$ (J in Hz)	0 _C
2	7.21(3)	121.0	6.63 (dd, 8.5, 2.5)	109.1	6.58 (dd, 8.5, 2.5)	107.9	6.52 (dd, 8.5, 2.5)	107.2	1.50 (5)	126.3
3 4	6.40 (s)	104.0	6.47 (d, 2.5)	101.1	6.48 (d, 2.5)	101.8	6.50 (d, 2.5)	100.9	6.92 (s)	102.3
4a 6	3.58 (dd, 11, 11)	155.1 66.5	3.63 (dd, 11, 11)	156.7 66.6	5.51 (s)	155.9 66.2	5.51 (s)	155.1 65.6		152.5 159.0
6 6a	4.20 (dd, 11, 5) 3.43 (ddd 11, 7, 5)	40.3	4.24 (dd, 11, 5) 3.46 (ddd 11, 7, 5)	40.2		107.5		106.6		102.1
11a	5.41 (d, 7)	78.3	5.44 (d, 7)	78.2		147.0		147.4		157.8
11b 10b		112.3		112.5		111.1		110.0 118.6		104.0 114.1
7 8	6.70 (s)	110.4 138.7	6.71 (s)	110.4 139.0	6.78 (s)	101.7 143.1	6.73 (s)	102.4 136.8	7.26 (s)	104.8 144.3
9		139.0		138.7		142.5		141.9	7.10 (a)	145.5
10 10a		148.2		148.4		149.8		145.2	7.19 (8)	98.9 148.8
OMe-at 1 OMe-at 3			3.79 (s)	55.4	3.80 (s)	55.7	3.80 (s)	55.4		
OMe-at 8										
Me at 8										
OH at 3 OH at 8							5.39 (brs)			
OH at 9										
maaiti	$\frac{35^a}{1}$	2		2						
position	$\sigma_{\rm H} (J {\rm in} {\rm Hz})$	120.2	7 86 (d. 9)	121.0						
2	(3)	126.0	7.08 (dd, 9, 2.5)	112.8						
5 4	6.87 (s)	158.5 102.3	7.16 (d, 2.5)	161.9 101.4						
4a		152.5		154.1						

Table 6. Continued

	35 ^{<i>a</i>}		36 ^{<i>a</i>}	
position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
6		158.8		158.2
6a		102.2		103.0
11a		157.8		157.6
11b		104.1		105.6
6b		113.5		113.2
7	7.12 (s)	102.0	7.17 (s)	102.0
8		143.9		144.0
9		142.8		143.1
10		112.3		112.4
10a		148.0		148.1
OMe-at 1				
OMe-at 3			3.89 (s)	55.8
OMe-at 8				
OMe-at 9				
Me at 8				
OH at 3				
OH at 8				
OH at 9				

^{*a*} Measured in CDCl₃. ^{*b*} Measured in acetone- d_6 . ^{*c*} Measured in DMSO- d_6 . ^{*d*} Isoprenyl groups were observed; one in **20**, **25–29**, **31–34**, **36** and two in **21–24**, **30**, **35**.

Table 7.	NMR	Spectroscopic	Data (4	400 MHz)	for Com	pounds 37–40
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	37 ^{<i>a</i>}		38 ^a		39 ^{<i>a</i>}		40 ^{<i>a</i>}	
position	$\delta_{\mathrm{H}} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\mathrm{H}} \left(J \text{ in Hz} \right)$	$\delta_{ m C}$	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{ m C}$	$\delta_{ m H}~(J~{ m in}~{ m Hz})$	$\delta_{\rm C}$
upper moiety								
2		154.6		153.5		154.5		153.4
3		120.4		120.3		120.4		120.3
4		196.4		196.3		196.3		196.3
4a		116.3		116.8		116.3		116.8
5	6.97 (s)	103.2	6.94 (s)	103.2	6.96 (s)	103.2	6.91 (s)	103.1
6		143.3		143.0		143.2		143.0
7		142.8		143.3		142.8		143.3
8		112.2		115.2		112.2		112.3
8a		148.4		148.4		148.3		148.4
1'		110.8		112.4		110.7		112.2
2'		154.9		156.8		154.9		156.8
3'	6.39 (s)	103.7	6.43 (d, 2)	102.7	6.40 (s)	103.7	6.42 (d, 2.5)	102.6
4'		158.2		163.0		158.2		163.0
5'		121.1	6.62 (dd, 8, 2)	106.8		121.0	6.61 (dd, 8.5, 2.5)	106.9
6'	7.32 (s)	131.2	7.59 (d, 8)	131.7	7.27 (s)	131.4	7.58 (d, 8.5)	131.6
OMe at 4'			3.79 (s)	55.7			3.78 (s)	
OH at 6								
OH at 7	8.35 (brs)							
OH at 2'	8.41 (brs)							
OH at 4'	8.46 (brs)							
lower moiety								
1	7.18 (d, 8.5)	129.7	7.17 (d, 8.5)	129.1	7.19 (s)	132.4	7.18 (s)	132.4
2	6.60 (d. 8.5)	110.1	6.61 (d. 8.5)	110.1		123.1	~ /	123.1
3		157.0		157.0		157.2		157.2
4		116.9		116.8	6.37 (s)	103.7	6.35 (s)	103.7
4a		155.4		155.4		155.6		155.7
6	3.16 (dd. 11, 11)	67.2	3.20 (dd. 10.5, 10.5)	67.3	3.21 (dd. 11, 10.5)	66.8	3.25 (dd. 10.5, 10.5)	66.9
6	3.95 (dd. 11, 5)				3.87 (dd. 11, 5)		3.89 (dd. 10.5, 5)	
6a	3.37 (dddd, 11, 7, 5, 1)	39.8		39.8	3.37 (dddd, 10.5, 7, 5, 1)	39.9	3.42 (dddd, 10.5, 7, 5, 1)	40.0
11a	5 48 (d. 7)	80.9		80.9	5 46 (d. 7)	80.4	5.50 (d. 7)	80.4
11b	5110 (a, 7)	112.4		112.2	5110 (d, 7)	112.0	5155 (d, 7)	112.1
10b		119.3		119.4		119.2		119.4
7	7.39 (d. 1)	128.1		128.0	7.38 (d. 1)	128.0	7 43 (d. 1)	127.9
8	(le, 1)	115.1		115.2	/180 (d, 1)	115.1	(iii) (iii, i)	115.2
9		164.5		164.5		164.5		164.5
10		11115		111.6		111.6		111.6
10a		164.3		164.3		164.4		164 5
OH at 3	8 35 (brs)	10115		101.5		101.1		101.5
OH at 9	13.42 (s)				13.41 (s)		13.43 (s)	

^{*a*} Measured in acetone- d_6 . Isoprenyl groups were observed; four in **37**, **38** and three in **38**, **40**.

11 (11.1 mg). Fraction 8C (26.6 mg) was subjected to PTLC [hexane-ethyl acetate (6:4)] to give 41 (10.2 mg). Fraction 8D (83.8 mg) [PTLC, hexane-ethyl acetate (6:4)] gave 37 (17.0 mg). Fraction 8E (71.7 mg) [PTLC, hexane-ethyl acetate (6:4)] yielded 39 (25.7 mg). Fraction 9A (54.3 mg) [PHPLC, Capcellpak phenyl 2×25 cm; H₂O-CH₃CN-trifluoroacetic acid (67.4:32.5:0.1), UV, 280 nm] gave 13 (19.9 mg) and 25 (24.0 mg).

Lespeflorin A₁ (1): colorless, amorphous solid; $[\alpha]_{D}^{23} - 43.0$ (*c* 0.74, MeOH); UV λ_{max} (log ϵ) 275 (4.10), 315 (3.76) nm; CD (MeOH) λ_{max} nm ([θ]) 240 (+22 736), 305 (-40 015), 337 (+28 192); ¹H NMR and ¹³C NMR, Table 1; HRFABMS *m*/*z* 405.2059 (calcd for C₂₆H₃₀O₄ – H, 405.2067).

Lespeflorin A₂ (2): colorless, amorphous solid; $[\alpha]_{D^3}^{D^3}$ –89.8 (*c* 0.23, MeOH); UV λ_{max} (log ϵ) 275 (3.95), 310 sh (3.76) nm; CD (MeOH)

Table 8. Melanin Synthesis Inhibitory Effects (IC₅₀ μ M) of Isolated Flavonoids

compd	IC ₅₀ (µM)	compd	IC ₅₀ (µM)
hydroquinone	2.20	isoflavone	
arbutin	103	47	29.2
euchrenone a ₆	1.02	benzofuran	
amorilin	1.21	18	10.8
amorisin	3.60	19	8.27
lespedezaflavanone H	7.80	pterocarpan	
•		20	3.73
flavanone		21	2.07
1	3.96	22	2.03
2	3.86	23	3.24
3	1.50	24	1.52
4	4.93	25	6.16
41	2.70	26	6.02
42	2.52	27	1.63
43	0.98	28	8.38
44	2.59	29	1.44
45	1.51	30	2.90
46	8.41	31	8.34
flavanonol		48	2.00
5	2.87	49	3.84
6	2.19	50	2.93
7	0.68	51	1.84
8	1.23	pterocarpene	
hydroxydihydrochalcone		32	27.0
9	>146	33	1.48
10	28.1	coumestan	
11	1.68	34	3.21
12	>135	35	28.3
13	>135	36	27.3
14	22.0	dimer	
15	3.78	37	39.6
isoflavanone		38	>52.8
16	1.47	39	37.7
isoflavene		40	>52.8
17	4.29		

 λ_{max} nm ([θ]) 300 (-12 882), 332 (+43 074); ¹H NMR and ¹³C NMR, Table 1; HRFABMS *m*/*z* 337.1432 (calcd for C₂₁H₂₂O₄ - H, 337.1440).

Lespeflorin A₃ (3): colorless, amorphous solid; $[\alpha]_{2^3}^{2^3} - 35.6$ (*c* 0.65, MeOH); UV λ_{max} (log ϵ) 279 (4.10), 321 sh (3.88) nm; CD (MeOH) λ_{max} nm ([θ]) 302 (-21 663), 340 (+24 995); ¹H NMR and ¹³C NMR, Table 1; HRFABMS *m*/*z* 457.2394 (calcd for C₃₀H₃₄O₄ – H, 457.2380).

Lespeflorin A₄ (4): colorless, amorphous solid; $[α]_D^{23} - 13.6$ (*c* 0.18, MeOH); UV $λ_{max}$ (log ε) 270 (4.76), 350 sh (4.02) nm; CD (MeOH) $λ_{max}$ nm ([θ]) 300 (-860); ¹H NMR and ¹³C NMR, Table 1; HRFABMS *m*/*z* 367.1209 (calcd for C₂₁H₂₀O₆ – H, 367.1182).

Lespeflorin B₁ (5): colorless, amorphous solid; $[\alpha]_D^{23} - 11.5$ (*c* 0.62, MeOH); UV λ_{max} (log ϵ) 279 (4.11), 325 (3.85) nm; CD (MeOH) λ_{max} nm ([θ]) 302 (-17 521), 333 (+12 302); ¹H NMR and ¹³C NMR, Table 2; HRFABMS *m*/*z* 407.1827 (calcd for C₂₅H₂₈O₅ – H, 407.1859).

Lespeflorin B₂ (6): colorless, amorphous solid; $[\alpha]_{D^3}^{23} - 2.5$ (*c* 1.04, MeOH); UV λ_{max} (log ϵ) 283 (4.10), 325 sh (3.76) nm; CD (MeOH) λ_{max} nm ([θ]) 307 (-23 203), 338 (+16 543); ¹H NMR and ¹³C NMR, Table 2; HRFABMS *m*/*z* 407.1882 (calcd for C₂₅H₂₈O₅ - H, 407.1859).

Lespeflorin B₃ (7): colorless, amorphous solid; $[\alpha]_{23}^{23} - 10.2$ (*c* 4.06, MeOH); UV λ_{max} (log ϵ) 283 (4.15), 319 sh (3.80) nm; CD (MeOH) λ_{max} nm ([θ]) 305 (-22 860), 338 (+23 495); ¹H NMR and ¹³C NMR, Table 2; HRFABMS *m*/*z* 475.2458 (calcd for C₃₀H₃₆O₅ – H, 475.2486).

Lespeflorin B₄ (8): colorless, amorphous solid; $[\alpha]_D^{23} - 27.9$ (*c* 0.29, MeOH); UV λ_{max} (log ϵ) 297 (4.12), 339 sh (3.73) nm; CD (MeOH) λ_{max} nm ([θ]) 298 (-22 628), 340 (+4764); ¹H NMR and ¹³C NMR, Table 2; HRFABMS *m*/*z* 491.2407 (calcd for C₃₀H₃₆O₆ – H, 491.2435).

Lespeflorin C₁ (9): colorless, amorphous solid; $[\alpha]_{D}^{23}$ +76.7 (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 282 (4.20), 330 (3.95) nm; CD (MeOH) λ_{max} nm ([θ]) 250 (-5702), 287 (+11 405), 330 (+13 230); ¹H NMR and ¹³C NMR, Table 3; HRFABMS *m/z* 341.1367 (calcd for C₂₀H₂₂O₅ – H, 341.1389).

Lespeflorin C₂ (10): colorless, amorphous solid; $[\alpha]_{D^3}^{D^3}$ +68.4 (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 274 (4.07), 316 (3.93) nm; CD (MeOH) λ_{max} nm ([θ]) 250 (-5936), 287 (+11 872), 330 (+13 772); ¹H NMR and ¹³C NMR, Table 3; HRFABMS *m*/*z* 355.1522 (calcd for C₂₁H₂₄O₅ – H, 355.1546).

Lespeflorin C₃ (11): colorless, amorphous solid; $[\alpha]_D^{23} - 2.9$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 274 (4.06), 316 (3.92) nm; CD (MeOH) λ_{max} nm ([θ]) 230 (-9203), 271 (+23 008), 297 (+10 123), 327 (-2956); ¹H NMR and ¹³C NMR, Table 3; HRFABMS *m*/*z* 355.1517 (calcd for C₂₁H₂₄O₅ - H, 355.1546).

Lespeflorin C₄ (12): colorless, amorphous solid; $[\alpha]_D^{23} - 7.0$ (*c* 0.56, MeOH); UV λ_{max} (log ϵ) 277 (4.18), 320 sh (3.87) nm; CD (MeOH) λ_{max} nm ([θ]) 235 (-9307), 270 (+23 268), 326 (-15 653); ¹H NMR and ¹³C NMR Table 3; HRFABMS *m*/*z* 369.1727 (calcd for C₂₂H₂₆O₅ - H, 369.1703).

Lespeflorin C₅ (13): colorless, amorphous solid; $[\alpha]_D^{23} + 72.5$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 287 (3.60), 331 (3.31) nm; CD (MeOH) λ_{max} nm ([θ]) 255 (-5406), 290 (+12 720), 332 (+9540); ¹H NMR and ¹³C NMR, Table 3; HRFABMS *m*/*z* 355.1533 (calcd for C₂₁H₂₄O₅ – H, 355.1546).

Lespeforin C₆ (14): colorless, amorphous solid; $[\alpha]_{D^2}^{D^2} + 89.5$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 287 (4.12), 330 sh (3.83) nm; CD (MeOH) λ_{max} nm ([θ]) 256 (-4280), 291 (+15 338), 332 (+9631); ¹H NMR and ¹³C NMR Table 3; HRFABMS *m*/*z* 409.2005 (calcd for C₂₅H₃₀O₅ - H, 409.2016).

Lespeflorin C₇ (**15**): colorless, amorphous solid; $[\alpha]_D^{23} - 43.3$ (*c* 0.66, MeOH); UV λ_{max} (log ϵ) 257 (4.39), 336 sh (3.85) nm; CD (MeOH) λ_{max} nm ([θ]) 288 (+10 821), 318 (+393), 340 (-4919); ¹H NMR and ¹³C NMR, Table 3; HRFABMS *m*/*z* 353.1376 (calcd for C₂₁H₂₂O₅ – H, 353.1389).

Lespeflorin D₁ (16): colorless, amorphous solid; $[\alpha]_{D}^{23} + 25.1$ (*c* 0.88, MeOH); UV λ_{max} (log ϵ) 282 (4.16) nm; CD (MeOH) λ_{max} nm ([θ]) 290 (-8649), 325 (+11 380); ¹H NMR and ¹³C NMR, Table 4; HRFABMS *m*/*z* 421.2010 (calcd for C₂₆H₃₀O₅ – H, 421.2016).

Lespeflorin E₁ (17): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 242 (4.14), 312 (4.13) nm; ¹H NMR and ¹³C NMR, Table 4; HRFABMS *m*/*z* 330.1095 (calcd for C₁₈H₁₈O₆, 330.1103).

Lespeflorin F₁ (18): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 312 (4.50), 326 (4.41) nm; ¹H NMR and ¹³C NMR, Table 5; HRFABMS *m*/*z* 315.0893 (calcd for C₁₇H₁₆O₆ – H, 315.0869).

Lespeflorin F₂ (19): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 312 (4.16) nm; ¹H NMR and ¹³C NMR, Table 5; HRFABMS *m*/*z* 369.1353 (calcd for C₂₁H₂₂O₆ – H, 369.1353).

Lespeflorin G₁ (20): colorless, amorphous solid; $[\alpha]_{D}^{23} - 110.0$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 300 (3.88), 341 (3.51) nm; CD (MeOH) λ_{max} nm ([θ]) 295 (+11 736); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 353.1370 (calcd for C₂₁H₂₂O₅ – H, 353.1389).

Lespeflorin G₂ (21): colorless, amorphous solid; $[α]_{D^3}^{23} - 50.2$ (*c* 1.35, MeOH); UV λ_{max} (log ϵ) 285 (3.98), 337 (3.33) nm; CD (MeOH) λ_{max} nm ([θ]) 295 (+4979); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 422.2060 (calcd for C₂₆H₃₀O₅, 422.2094).

Lespeflorin G₃ (22). colorless, amorphous solid; $[α]_{D}^{23} - 178.1$ (*c* 1.55, MeOH); UV λ_{max} (log ϵ) 287 (3.94), 340 (3.84) nm; CD (MeOH) λ_{max} nm ([θ]) 288 (+3211); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 452.2180 (calcd for C₂₇H₃₂O₆, 452.2200).

Lespeflorin G₄ (23). colorless, amorphous solid; $[α]_{D^3}^{2^3} - 177.1$ (*c* 0.50, MeOH); UV λ_{max} (log ϵ) 290 (4.04), 345 (3.56) nm; CD (MeOH) λ_{max} nm ([θ]) 292 (+15 607); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 406.2166 (calcd for C₂₆H₃₀O₄, 406.2145).

Lespeflorin G₅ (24): colorless, amorphous solid; $[α]_{D}^{23}$ –55.6 (*c* 1.70, MeOH); UV $λ_{max}$ (log ϵ) 278 (3.83), 326 (3.36) nm; CD (MeOH) $λ_{max}$ nm ([θ]) 286 (+13160); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 406.2176 (calcd for C₂₆H₃₀O, 406.2145).

Lespeflorin G₆ (25): colorless, amorphous solid; $[\alpha]_{D}^{23} - 118.1$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 300 (3.75), 340 (3.19) nm; CD (MeOH) λ_{max} nm ([θ]) 290 (+10 624); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 353.1373 (calcd for C₂₁H₂₂O₅ – H, 353.1389).

Lespeflorin G₇ (26): colorless, amorphous solid; $[\alpha]_{D^2}^{23}$ –84.7 (*c* 0.18, MeOH); UV λ_{max} (log ϵ) 281 (3.87), 330 (3.70) nm; CD (MeOH) λ_{max} nm ([θ]) 289 (+5407); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 354.1482 (calcd for C₂₁H₂₂O₅, 354.1468).

Lespeflorin G₈ (27): colorless, amorphous solid; $[\alpha]_{D}^{23} - 121.8$ (*c* 0.74, MeOH); UV λ_{max} (log ϵ) 300 (3.66), 340 sh (3.30) nm; CD (MeOH) λ_{max} nm ([θ]) 290 (+5647); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 367.1550 (calcd for C₂₂H₂₄O₅ – H, 367.1546).

Lespeflorin G₉ (28): colorless, amorphous solid; $[\alpha]_{D}^{23} - 109.3$ (*c* 0.77, MeOH); UV λ_{max} (log ϵ) 279 (3.91), 333 (3.58) nm; CD (MeOH) λ_{max} nm ([θ]) 289 (+8516); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 368.1616 (calcd for C₂₂H₂₄O₅, 368.1624).

Lespeflorin G₁₀ (29): colorless, amorphous solid; $[\alpha]_D^{23} -91.7$ (*c* 0.42, MeOH); UV λ_{max} (log ϵ) 282 (3.94), 334 (3.84) nm; CD (MeOH) λ_{max} nm ([θ]) 287 (+7546); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 338.1534 (calcd for C₂₁H₂₂O₄, 338.1519).

Lespeflorin G₁₁ (**30**): colorless, amorphous solid; $[α]_D^{23} - 233.7$ (*c* 0.72, MeOH); UV λ_{max} (log ϵ) 271 (4.02), 344 (3.66) nm; CD (MeOH) λ_{max} nm ([θ]) 289 (+3450); ¹H NMR and ¹³C NMR, Table 1; HRFABMS *m*/*z* 406.1747 (calcd for C₂₅H₂₆O₅, 406.1781).

Lespeflorin G₁₂ (**31**): colorless, amorphous solid; $[\alpha]_D^{23} - 110.0$ (*c* 0.45, MeOH); UV λ_{max} (log ϵ) 271 (3.99), 341 (3.66) nm; CD (MeOH) λ_{max} nm ([θ]) 284 (+8200); ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 351.1247 (calcd for C₂₁H₂₀O₅ – H, 351.1233).

Lespeflorin H₁ (32): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 339 (4.32), 355 (4.29) nm; ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 351.1247 (calcd for C₂₁H₂₀O₅ – H, 351.1233).

Lespeflorin H₂ (33): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 358 (4.01), 367 sh (3.99) nm; ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 349.1082 (calcd for C₂₁H₁₈O₅ – H, 349.1076).

Lespeflorin I₁ (34): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 353 (4.24), 366 sh (4.20) nm; ¹H NMR and ¹³C NMR, Table 6; HRFABMS *m*/*z* 351.0898 (calcd for C₂₀H₁₆O₆ – H, 351.0869).

Lespeflorin I₂ (35): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 355 (4.22), 367 sh (4.18) nm; ¹H NMR and ¹³C NMR Table 6; HRFABMS *m*/*z* 419.1494 (calcd for C₂₅H₂₄O₆ – H, 419.1495).

Lespeflorin I₃ (36): colorless, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 352 (4.26), 363 sh (4.23) nm; ¹H NMR and ¹³C NMR Table 6; HRFABMS *m*/*z* 365.1028 (calcd for C₂₁H₁₈O₆ – H, 365.1025).

Lespeflorin J₁ (37): colorless, amorphous solid; $[\alpha]_D^{23} - 17.8$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 315 (439) nm; CD (MeOH) λ_{max} nm ([θ]) 232 (-48 741), 254 (+8124), 272 (-34 931), 329 (-2464), 394 (+14 622); ¹H NMR and ¹³C NMR, Table 7; HRFABMS *m/z* 811.3492 (calcd for C₅₀H₅₂O₁₀ - H, 811.3483).

Lespeflorin J₂ (38): colorless, amorphous solid; $[\alpha]_D^{23} - 9.0$ (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 314 (4.39) nm; CD (MeOH) λ_{max} nm ([θ]) 231 (-31 862), 254 (+9559), 270 (-25 489), 329 (-2218), 385 (+10 833); ¹H NMR and ¹³C NMR, Table 7; HRFABMS *m/z* 757.3016 (calcd for C₄₆H₄₆O₁₀ - H, 757.3014).

Lespeflorin J₃ (39): colorless, amorphous solid; $[\alpha]_{D^3}^{23}$ –67.4 (*c* 1.00, MeOH); UV λ_{max} (log ϵ) 320 (4.42) nm; CD (MeOH) λ_{max} nm ([θ]) 233 (-40 618), 255 (+1310), 272 (-32 756), 325 (-2271), 393 (+13 103); ¹H NMR and ¹³C NMR, Table 7; HRFABMS *m/z* 811.3492 (calcd for C₅₀H₅₂O₁₀ – H, 811.3483).

Lespeflorin J₄ (40): colorless, amorphous solid; $[\alpha]_{D}^{23} - 55.0$ (*c* 1.00, MeOH); UV, λ_{max} (log ϵ) 316 (4.43) nm; CD (MeOH) λ_{max} nm ([θ]) 233 (-41 665), 255 (+4167), 270 (-25 833), 329 (-2321), 390 (+12 500); ¹H NMR and ¹³C NMR, Table 7; HRFABMS *m/z* 757.3016 (calcd for C₄₆H₄₆O₁₀ - H, 757.3014).

Assay of Melanin Synthesis in NHEM. NHEM were purchased from Kurabo Industries Ltd. (Osaka, Japan) and were maintained in M254 calcium-free medium supplemented with human melanocyte growth supplement (HMGS), 50 µg/mL streptomycin, and 50 U/mL penicillin. NHEM were seeded in a 24-well culture plate. After incubation for 24 h, the medium was exchanged for the sample compound dissolved in DMSO and 0.25 µCi ¹⁴C-thiouracil (Amersham, Freiburg, Germany). After incubation for 72 h, the cells were washed twice with phosphate-buffered saline and lysed with trichloroacetic acid (TCA). After centrifugation, the pellets were washed twice with 10% TCA. The pellets were mixed with scintillation fluid, and the incorporated radioactivity was determined with a liquid-scintillation counter (LSC6100; ALOKA, Tokyo, Japan). Cell viability was examined by a colorimetric assay using a WST-8 cell-counting kit (Dojindo, Kumamoto, Japan) according to the manufacturer's protocol. A WST-8 [2-(2-methoxy-4-nitrophenyl)-3-(4-nitrophenyl)-5-(2,4-disulfophenyl)-2*H*-tetrazolium, monosodium salt] solution was added after the 72 h treatment and then incubated for another 3 h at 37 °C. The absorbance of each well was measured at 450 nm with a reference wavelength at 630 nm. The mean absorbance of control wells (cells without compound) represented 100% cell viability. Viability of compound-treated cells was determined in triplicate and related to the absorbance of control cells. The 50% inhibition concentration (IC₅₀) value of melanin synthesis was determined using SAS software version 9.1.3 (SAS Institute Inc., Cary, NC).

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Supporting Information Available: Structures of known compounds and all NMR data (isopropyl residue, ROE, and HMBC). This material is available free of charge via the Internet at http://pubs.acs.org.

References and Notes

- (1) Katagiri, T.; Hongo, M.; Miyase, T. Jpn. Kokai Tokkyo KohoJP2007186441, 2007.
- (2) Bhalla, V. K.; Nayak, U. R. Tetrahedron Lett. 1968, 2401-2406.
- (3) Kyogoku, K.; Hatayama, K.; Suzuki, K.; Yokomori, S.; Maejima, K.; Komatsu, M. Chem. Pharm. Bull. 1973, 21, 1777–1782.
- (4) Iinuma, M.; Ohyama, M.; Tanaka, T. Phytochemistry 1995, 38, 539– 543.
- (5) (a) Dawidar, A. M.; Jakupovic, J.; Abdel-Mogib, M.; Mashaly, I. A. *Phytochemistry* 1994, 36, 803–806. (b) Schütz, B. A.; Wrig, A. D.; Rali, T.; Sticher, O. *Phytochemistry* 1995, 40, 1273–1277.
- (6) Mizuno, M.; Tamura, K.; Tanaka, T.; Iinuma, M. Phytochemistry 1988, 27, 1831–1834.
- (7) (a) Bhanumati, S.; Chhabra, S. C.; Gupta, S. R.; Krishnamoorthy, V. *Phytochemistry* **1978**, *17*, 2045. (b) Fomum, Z. T.; Ayafor, J. F.; Wandji, J. *Phytochemistry* **1985**, *24*, 3075–3076.
- (8) Miyase, T.; Ueno, A.; Noro, T.; Fukushima, S. Chem. Pharm. Bull. 1980, 28, 1172–1177.
- (9) Iinuma, M.; Tanaka, T.; Mizuno, M.; Yamamoto, H.; Kobayashi, Y.; Yonemori, S. *Chem. Pharm. Bull.* **1992**, 40, 2749–2752.
- (10) Mitscher, L. A.; Okwute, S. K.; Gollapudi, S. R.; Drake, S. R.; Avona, E. *Phytochemistry* **1988**, *27*, 3449–3452.
- (11) Tanaka, H.; Tanaka, T.; Etho, H. Phytochemistry 1998, 47, 475-477.
- (12) Andersen, Ø. M.; Markham, K. R. Flavonoids; Taylor & Francis: New York, 2006; p 56.
- (13) Schütz, B. A.; Wright, A. D.; Pali, T.; Sticher, O. *Phytochemistry* **1995**, 40, 1273–1277.
- (14) Slade, D.; Ferreira, D.; Marais, J. P. J. Phytochemistry 2005, 66, 2177– 2215.
- (15) Harborne, J. B. *The Flavonoids*; Chapman & Hall: New York, 1986; p 462.
- (16) Harborne, J. B. *The Flavonoids*; Chapman & Hall: New York, 1986; p 421.
- (17) Fukai, T.; Sheng, C.-B.; Horikoshi, T.; Nomura, T. *Phytochemistry* **1996**, *43*, 1119–1124.
- (18) Augustyn, J. A. N.; Bezuidenhoudt, B. C. B.; Swanepoel, A.; Ferreira, D. *Tetrahedron* **1990**, *46*, 4429–4442.
- (19) Miyase, T.; Sano, M.; Yoshino, K.; Nonaka, K. *Phytochemistry* **1999**, *52*, 311–319.
- (20) Miyase, T.; Sano, M.; Nakai, H.; Muraoka, M.; Nakazawa, M.; Suzuki, M.; Yoshino, K.; Nishihara, Y. *Phytochemistry* **1999**, *52*, 303–310.
- (21) Harborne, J. B. *The Flavonoids*; Chapman & Hall: New York, 1986; p 167.
- (22) (a) Napolitano, A.; Palumbo, A.; d'Ischia, M.; Prota, G. J. Med. Chem. 1996, 39, 5192–5201. (b) Yada, Y.; Higuchi, K.; Imokawa, G. J. Biol. Chem. 1991, 266, 18352–18357. (c) Whittaker, J. R. J. Biol. Chem. 1971, 246, 6217–6226.

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